# Exercise 1.1: Fundamental Forces

What are the known fundamental interactions? Consider the following objects and discuss which interaction is essential for both its interaction and structure:

# • Galaxy:

- Essential Interaction: Gravitational
- Explanation: Gravity binds stars and other matter in a galaxy, determining its structure and dynamics over vast distances.

# • Solar System:

- Essential Interaction: Gravitational
- **Explanation:** The gravitational force from the Sun governs the orbits of the planets and maintains the structure of the solar system.

#### • Planet:

- Essential Interaction: Gravitational
- **Explanation:** Gravity holds a planet's material together, giving it shape and enabling it to maintain an atmosphere.

## • Basketball:

- Essential Interaction: Gravitational
- Explanation: Gravity influences the basketball's motion, especially when thrown or dropped, affecting its trajectory and interactions with the ground.

#### • Bacterium:

- Essential Interaction: Electromagnetic
- Explanation: Electromagnetic forces govern biochemical processes and molecular interactions within the bacterium.

# • Molecule:

- Essential Interaction: Electromagnetic
- Explanation: Electromagnetic interactions (chemical bonds) define the structure and stability of molecules.

#### • Atom:

- Essential Interaction: Electromagnetic (for electrons), Strong (for the nucleus)
- Explanation: Electromagnetic forces hold electrons in orbit around the nucleus, while the strong force binds protons and neutrons together in the nucleus.

## • Atomic Nucleus:

- Essential Interaction: Strong
- Explanation: The strong force is essential for binding protons and neutrons together in the nucleus, overcoming their electromagnetic repulsion.

#### • Proton:

- Essential Interaction: Strong
- Explanation: The strong force binds quarks together to form protons, defining their structure and interactions.

# • Quark:

- Essential Interaction: Strong
- Explanation: Quarks interact primarily through the strong force, which governs their behavior and holds them together to form protons and neutrons.

## Exercise 1.2: Natural Units

In the course, so-called natural units are used. That means we set  $c = \hbar = k_B = 1$ . As the name suggests, this does not change the physics, but only the units of our quantities. This exercise is about determining conversion factors between natural units and SI units.

- (a) What is a second in  $GeV^{-1}$ ?
  - **Solution:**  $1\text{sc}/(\hbar c) = 1.519 \cdot 10^{24}/\text{GeV}$  oder t = 1/GeV in natural units means  $t = 6.582 \cdot 10^{-25} \text{s}$  in SI units.
- (b) What is a meter in  $GeV^{-1}$ ?
  - **Solution:**  $1\text{m}/(\hbar c) = 10^{15} \text{ fm}/\hbar c = 5.068 \cdot 10^{15}/\text{GeV}$  or L = 1/GeV in natural units means  $0.1973 \text{ fm} = 1.973 \cdot 10^{-16} \text{m}$
- (c) What is the unit of momentum in the SI system and in natural units? Determine the conversion factor.
  - **Solution:**  $p = 1 \text{GeV}/c = 5.344 \cdot 10^{-19} \text{kgm/s}$  which is written as p = 1 GeV in natural units (since c = 1)
- (d) What is the unit of temperature in natural units? What is a Kelvin in this unit?
  - **Solution:**  $1 \text{K} k_{\text{B}} = 8.617 \cdot 10^{-14} \text{ GeV} = 8.617 \cdot 10^{-5} \text{ eV}$  or T = 1 GeV in natural units means  $T = 1.6 \cdot 10^{13} \text{K}$  in SI units.
- (f) What is a second in fm? Use the results of parts (a) and (e).
  - **Solution:**  $1 \text{ s} c = 2.998 \cdot 10^{23} \text{fm or } 1 \text{ s} c/(\hbar c) = 1.519 \cdot 10^{24} / \text{GeV}; t = 1/\text{GeV}$  in natural units thus means  $t = 6.582 \cdot 10^{-25} \text{s}$  in SI units.

# Exercise 1.3: Form factor and charge radius

The form factor for the scattering of an electron with a nucleus is given by

$$F(\vec{q}) = \frac{1}{Ze} \int_{\mathbb{R}^3} d^3 r \exp(i\vec{r} \cdot \vec{q}) \rho(\vec{r}),$$

where  $\rho$  is the charge density of the nucleus, and  $\vec{q}$  is the momentum transfer in the scattering. Assume a spherically symmetric charge distribution, i.e.,  $\rho(\vec{r}) = \rho(r)$  (with  $r = |\vec{r}|$ ) and that the typical scale of the nucleus's size  $R_{\rm n}$  is such that one can assume  $qR_{\rm nucl} \ll 1$  within the integral. Show that for these small momentum transfers, the form factor is given by

$$F(\vec{q}) = F(|\vec{q}|) = 1 - \frac{1}{6}\vec{q}^{2}\langle r^{2}\rangle$$

with

$$\langle r^2 \rangle = \int_{R^3} \mathrm{d}^3 r r^2 \rho(r),$$

i.e., from measuring the form factor you can deduce the "root-mean-square charge radius"  $R_{\rm rms}=\sqrt{\langle r^2\rangle}\simeq R_{\rm nucl}$ .

**Solution:** We can expand the exponential under the integral up to 2nd order. In spherical coordinates we have  $\vec{r} \cdot \vec{q} = rq \cos \vartheta$  (taking the polar axis in direction of  $\vec{q}$ ). Then

$$F(\vec{q}) \simeq \frac{1}{Ze} \int_0^\infty dr \int_0^\pi d\vartheta \int_0^{2\pi} d\varphi r^2 \sin\vartheta \left( 1 + irq\cos\vartheta - \frac{1}{2}r^2q^2\cos^2\vartheta \right) \rho(r)$$
$$= 1 - \frac{1}{6Ze} \int_0^\infty dr (4\pi r^2) r^2 \rho(r) = 1 - \frac{1}{6}q^2 \langle r^2 \rangle.$$

The integral over  $\vartheta$  is done by substitution of  $u = \cos \vartheta$ ,  $du = -d\vartheta \sin \vartheta$ . Also we used that the total charge is

$$\int_{\mathbb{R}^3} d^3r \rho(r) = \int_0^\infty dr 4\pi r^2 \rho(r) = Ze.$$